



Asymptotic Growth in Open Chemical Reaction Networks: Dynamics and Thermodynamics

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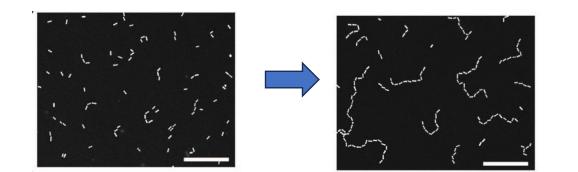
Phys. Rev. E, 109, 064153 (2024)

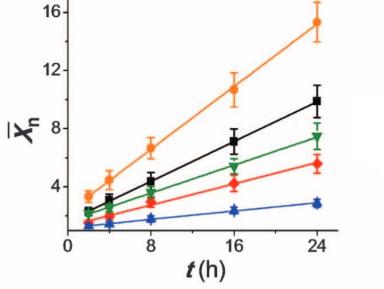
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Outline

- Motivation and Problem Statement
- Formalism: Dynamics and Thermodynamics of open CRNs
- Results
- Prospects

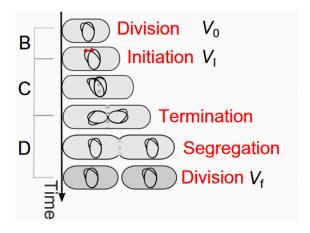
Growth is everywhere



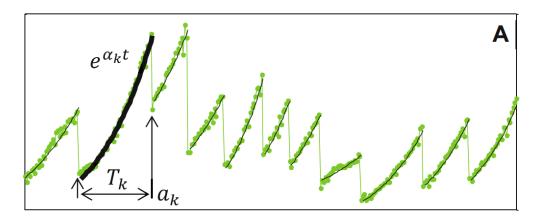


Science, 329, 197 (2010)

Polymerization



E-coli cell cycle *Sci. Adv.,* **4**,eaau3324(2018)



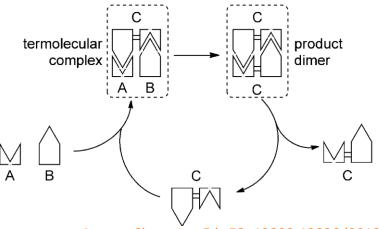
Protein count

Eur. Phys. J. E 38, 102 (2015)

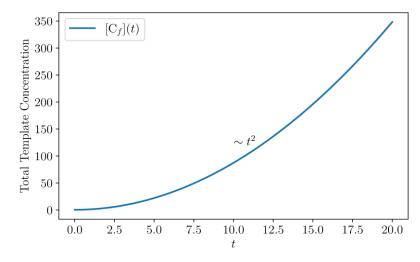
Concentration Growth in the Literature

Non enzymatic Template Replication

 $A + B + C \xrightarrow[-1]{-1} ABC$ $ABC \xrightarrow{+2} C_2$ $C_2 \xrightarrow[-3]{+3} 2C$



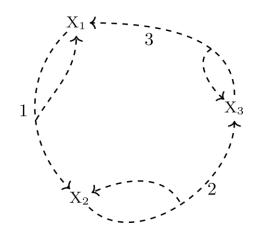


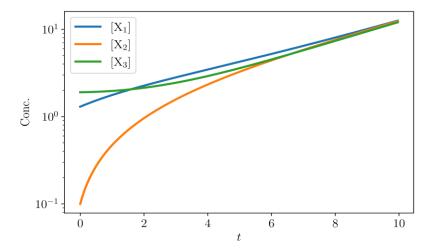


Hinshelwood Cycle

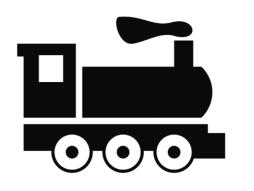
$$\begin{array}{c} X_1 \xrightarrow{1} X_1 + X_2 \\ X_2 \xrightarrow{2} X_2 + X_3 \\ X_3 \xrightarrow{3} X_3 + X_1 \end{array}$$

Phys. Rev. Lett. 113, 028101 (2014)

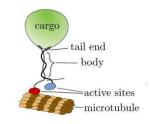


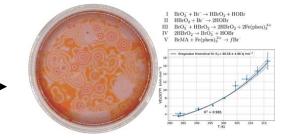


Nonequilibrium Thermodynamics



Phys. Rev. E 92, 042133 (2015)



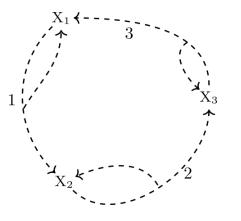


Macroscale: Equilibrium Thermodynamics Mesoscale: Stochastic Thermodynamics

Rep. Prog. Phys. **75** 126001(2012) *Physica A*, **418**, 6-16 (2015) Macroscopic: Nonequilibrium Thermodynamics

> J. Chem. Phys. 141, 024117 (2014) Phys. Rev. X 6, 041064 (2016) J. Chem. Phys. 154, 094114 (2021)

Problem – Irreversible and coarse-grained reactions!



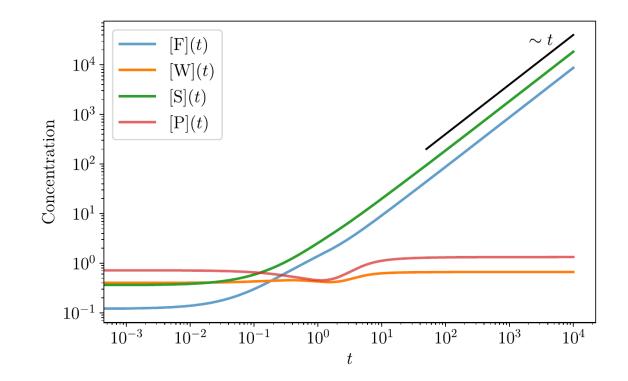
Question: Is concentration growth possible in CRNs with all reversible reactions?

Growth in open CRNs

 $E = P_{2}$ P_{3} E_{W} P_{3} P_{5} P E_{W} P_{5} P

Minimal Model of a Metabolic Network

J. Chem. Phys. 156, 014116 (2022)



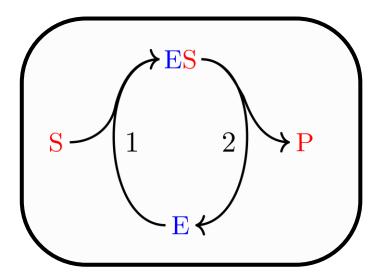
Key Questions:

- When does growth* occur?
- What is the dynamics of the growing state?
- Can we estimate the **cost** of growth?

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Topology of CRNs: Setup



$$\boldsymbol{\alpha} = (E, ES, S, P)$$

$$\mathbb{S} = \mathbb{E} \begin{bmatrix} 1 & 2 \\ -1 & 1 \\ 1 & -1 \\ -1 & 0 \\ P \end{bmatrix}$$

$$\boldsymbol{\ell}^{\mathrm{S}} = \begin{pmatrix} \mathrm{E} & \mathrm{ES} & \mathrm{S} & \mathrm{P} \\ 0 & 1 & 1 & 1 \end{pmatrix}$$
$$\boldsymbol{\ell}^{\mathrm{E}} = \begin{pmatrix} \mathrm{E} & \mathrm{ES} & \mathrm{S} & \mathrm{P} \\ 1 & 1 & 0 & 0 \end{pmatrix}$$

Michealis-Menten CRN

$$oldsymbol{
u}_{+
ho}\cdot lpha \stackrel{+
ho}{\underset{-
ho}{\longleftarrow}} oldsymbol{
u}_{-
ho}\cdot lpha$$

Formal reaction

$$\mathbb{S}_{
ho} = oldsymbol{
u}_{-
ho} - oldsymbol{
u}_{+
ho}$$

Stoichiometric Matrix

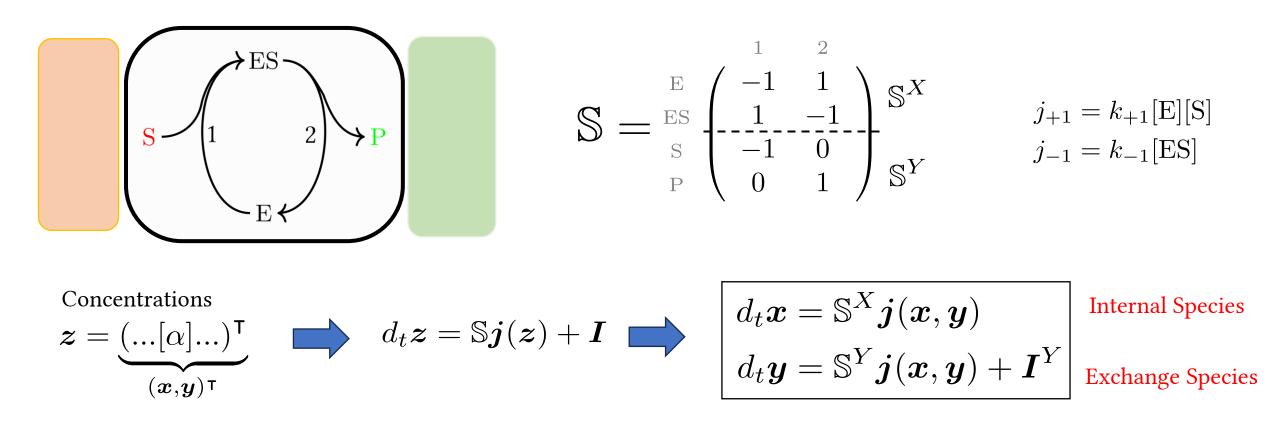
 $\boldsymbol{\ell}^{\lambda} \cdot \mathbb{S} = 0$

Conservation Laws

 $\boldsymbol{\ell}^m > 0$

Mass Conservation Law

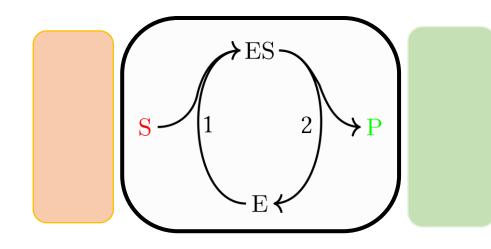
Dynamics of CRNs : Reactions



Reaction currents

 $j_{\rho}(\boldsymbol{z}(t)) = j_{+\rho}(\boldsymbol{z}(t)) - j_{-\rho}(\boldsymbol{z}(t)) \implies j_{\pm\rho}(\boldsymbol{z}(t)) = k_{\pm\rho}\boldsymbol{z}^{\boldsymbol{\nu}_{\pm\rho}}$ Ideal Dilute Solutions

Dynamics of CRNs : Reactions II



$$\mathbb{S} = \frac{\mathbb{E}}{\mathbb{S}} \left(\begin{array}{ccc} -1 & 1 \\ 1 & -1 \\ -1 & 0 \\ 0 & 1 \end{array} \right)$$

$$\boldsymbol{\ell}^{\mathrm{S}} = \begin{pmatrix} \mathrm{E} & \mathrm{ES} & \mathrm{S} & \mathrm{P} \\ 0 & 1 & 1 & 1 \end{pmatrix}$$
$$d_t L^{\mathrm{S}} = I_{\mathrm{S}} + I_{\mathrm{P}} > 0$$

$$\boldsymbol{\ell}^{\mathrm{E}}=\left(egin{array}{cccc} {}^{\mathrm{E}} {}^{\mathrm{E}} {}^{\mathrm{S}} {}^{\mathrm{S}} {}^{\mathrm{P}} \ 1 {}^{\mathrm{I}} {}^{\mathrm{I}} {}^{\mathrm{O}} {}^{\mathrm{O}} {}^{\mathrm{O}} \end{array}
ight)$$

 $d_{t} L^{\mathrm{E}}=0$

Exchange Mechanisms

Concentration control

$$d_t \boldsymbol{x} = \mathbb{S}^X \boldsymbol{j} \left(\boldsymbol{x}(t), \boldsymbol{y}(0) \right)$$

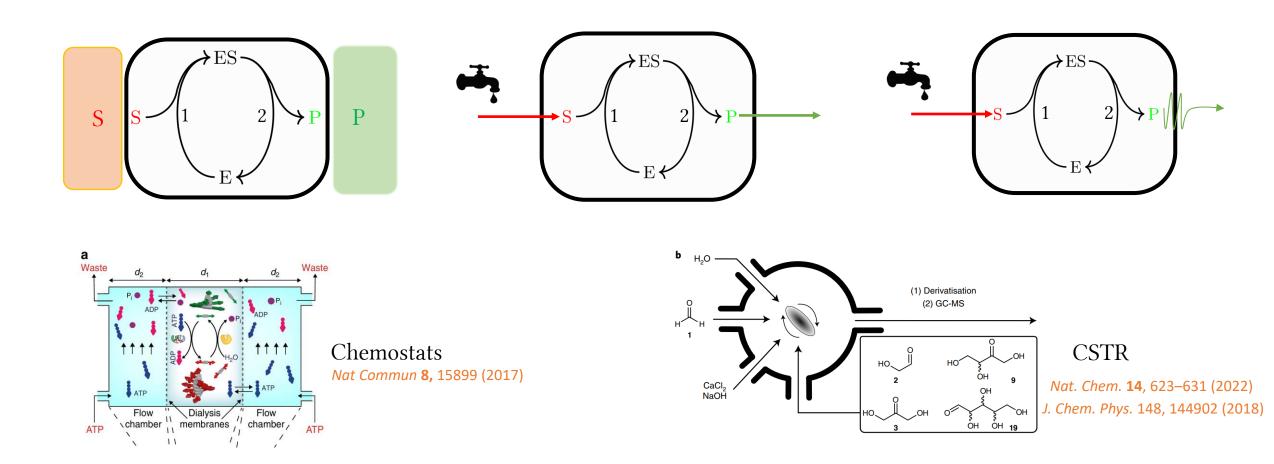
$$d_t \boldsymbol{y} = \mathbb{S}^Y \boldsymbol{j} \left(\boldsymbol{x}(t), \boldsymbol{y}(0) \right) + \boldsymbol{I}^Y = 0$$

Flux control

$$egin{aligned} &d_t oldsymbol{x} = \mathbb{S}^X oldsymbol{j} \left(oldsymbol{x}(t), oldsymbol{y}(t)
ight) \ &d_t oldsymbol{y} = \mathbb{S}^Y oldsymbol{j} \left(oldsymbol{x}(t), oldsymbol{y}(t)
ight) + \widetilde{oldsymbol{I}} \end{aligned}$$

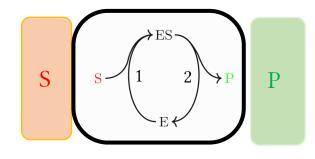
Mixed control

$$egin{aligned} &d_t oldsymbol{x} = \mathbb{S}^X oldsymbol{j} \left(oldsymbol{x}(t), oldsymbol{y}(t)
ight) \ &d_t oldsymbol{y} = \mathbb{S}^Y oldsymbol{j} \left(oldsymbol{x}(t), oldsymbol{y}(t)
ight) - ilde{\mathbb{D}} oldsymbol{y} + ilde{oldsymbol{I}} \end{aligned}$$



Thermodynamics

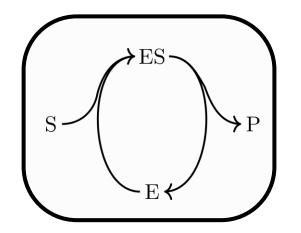
Chemical Potentials $\mu(z) = \mu^0 + RT \ln(z)$ Local Detailed $\mu \cdot \mathbb{S}_{\rho} = -RT \ln \left(\frac{j_{+\rho}(z)}{j_{-\rho}(z)} \right)$ $T\dot{\Sigma} = \dot{w}_c - d_t G \ge 0$ $T\dot{\Sigma} = RT \sum_{\rho} j_{\rho} \ln \left(\frac{j_{+\rho}}{j_{-\rho}} \right) \ge 0$ $\dot{w}_c = \sum_{\substack{\alpha \in Y \\ \text{Chemical Work rate}}} \mu_{\alpha} I_{\alpha}^Y$ $G(z) = \sum_{\substack{\alpha \in Z \\ \text{Gibbs Free Energy}}} (\mu_{\alpha} - RT) [\alpha]$

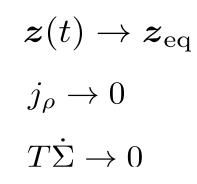


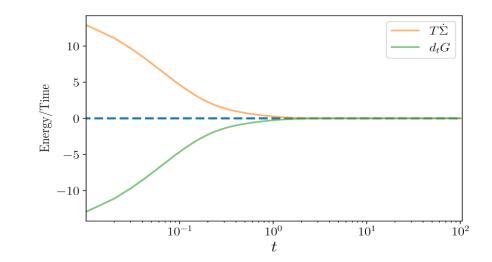
$$\dot{w}_{\rm c} = \mu_{\rm S} I_{\rm S} + \mu_{\rm P} I_{\rm P}$$

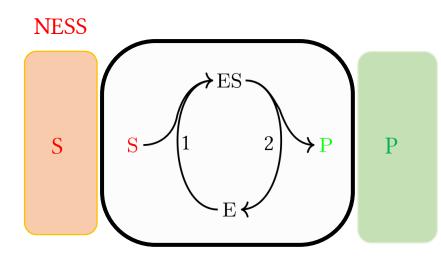
Thermodynamics: Equilibria vs NESS

Equilibrium



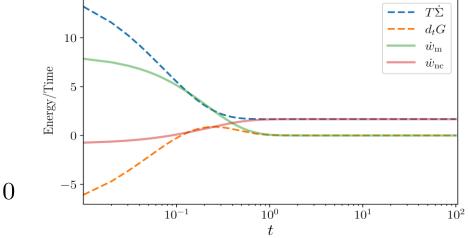




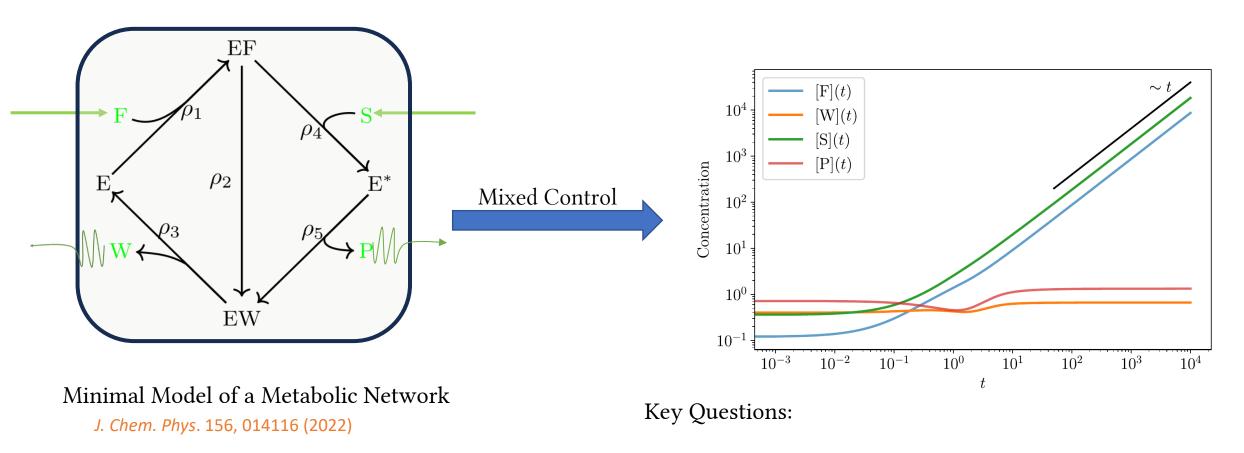


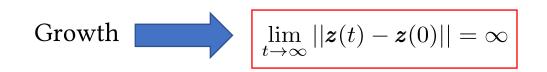
$$oldsymbol{z}(t)
ightarrow oldsymbol{z}_{
m ss}$$

 $j_{
ho}
ightarrow ar{j}
eq 0$
 $T\dot{\Sigma}, \dot{w}_c
ightarrow (\mu_{
m P} - \mu_{
m S}) ar{j} > 0$



Growth with reversible reactions





- When does growth* occur?
- What is the dynamics of the growing state?
- Can we estimate the **cost** of growth?

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Results: When?

Criteria for Growth:

Growth
$$\iff L^m(t) \to \infty \iff G(\boldsymbol{z}) \to \infty$$

Closed CRN:

$$d_t L^m = 0$$

No Growth

CSTR $d_t L^m = -\ell^m \cdot \mathbb{D} \boldsymbol{z} + \sum_{\alpha} \ell^m_{\alpha} I_{\alpha} = -\epsilon L^m + \sum_{\alpha} \ell^m_{\alpha} I_{\alpha} \qquad \text{No Growth}$

Flux Control

$$d_t L^m = \sum_{\alpha} \ell^m_{\alpha} I_{\alpha} \ge 0$$
 Growth!

Results: Systematic Analysis

	Flux Control	Mixed Control	Conc. Control
Dynamically Linear	Growth	No Growth	No Growth
Dynamically Nonlinear	Growth	Growth	No Growth*

Dynamically Linear: Unimolecular, Pseudo-unimolecular CRNs

Dynamically Nonlinear: Multimolecular CRNs

Exchange Mechanisms

Concentration control

$$d_t \boldsymbol{x} = \mathbb{S}^X \boldsymbol{j} \left(\boldsymbol{x}(t), \boldsymbol{y}(0) \right)$$

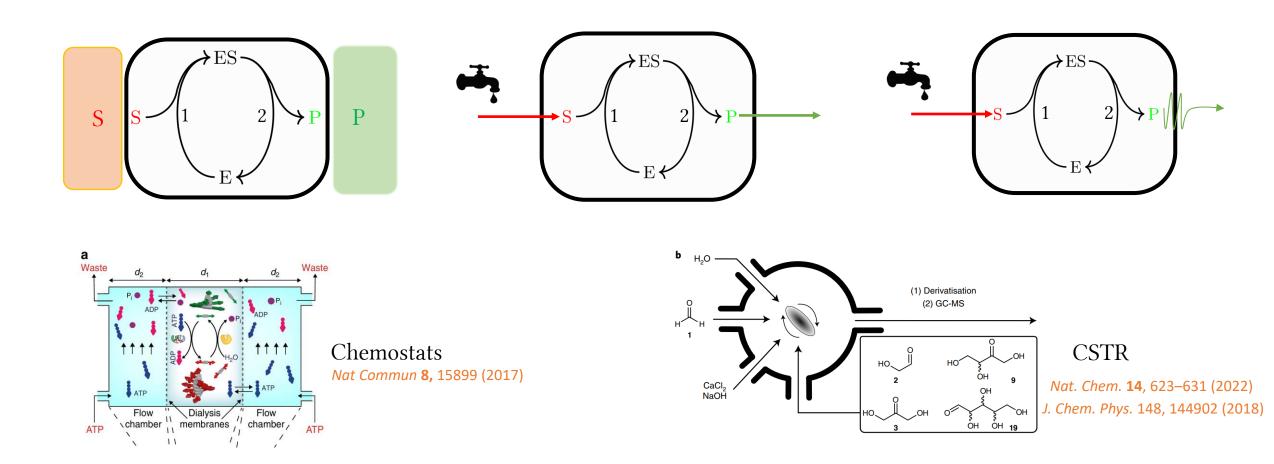
$$d_t \boldsymbol{y} = \mathbb{S}^Y \boldsymbol{j} \left(\boldsymbol{x}(t), \boldsymbol{y}(0) \right) + \boldsymbol{I}^Y = 0$$

Flux control

$$egin{aligned} &d_t oldsymbol{x} = \mathbb{S}^X oldsymbol{j} \left(oldsymbol{x}(t), oldsymbol{y}(t)
ight) \ &d_t oldsymbol{y} = \mathbb{S}^Y oldsymbol{j} \left(oldsymbol{x}(t), oldsymbol{y}(t)
ight) + \widetilde{oldsymbol{I}} \end{aligned}$$

Mixed control

$$egin{aligned} &d_t oldsymbol{x} = \mathbb{S}^X oldsymbol{j} \left(oldsymbol{x}(t), oldsymbol{y}(t)
ight) \ &d_t oldsymbol{y} = \mathbb{S}^Y oldsymbol{j} \left(oldsymbol{x}(t), oldsymbol{y}(t)
ight) - ilde{\mathbb{D}} oldsymbol{y} + ilde{oldsymbol{I}} \end{aligned}$$



Unimolecular CRNs

$$X_{1} \xrightarrow{+1}_{-1} X_{2}$$
$$X_{2} \xrightarrow{+2}_{-2} X_{3}$$
$$X_{3} \xrightarrow{+4}_{-3} X_{4}$$
$$X_{4} \xrightarrow{+4}_{-4} X_{1}$$

$$\boldsymbol{\ell^m} = (1, 1, 1...)^{\mathsf{T}}$$

Mass Conservation Law

$$\begin{array}{c} W \boldsymbol{\pi}^{\mathrm{eq}} = 0 \\ d_t \boldsymbol{z} = W \boldsymbol{z} \\ \text{Closed dynamics} \\ \lambda(W) \leq 0 \end{array} \boldsymbol{\lambda}(W) \leq 0 \end{array} \boldsymbol{\lambda}(W) \leq 0$$

Universal form for open dynamics

$$d_t \boldsymbol{z} = (\mathbb{W} - \mathbb{D})\boldsymbol{z} + \boldsymbol{I}$$



Unimolecular CRNs

Mass Conservation

	Flux Control	Mixed Control	Conc. Control
Dynamically Linear	Growth	No Growth	No Growth
Dynamically Nonlinear	Growth	Growb	No Growth*

 $\mathcal{L}^{m} = (1, 1, 1...)^{\mathsf{T}}$ $X_{1} \stackrel{+1}{\underbrace{\overline{1}}_{-1}} X_{2}$ $X_{2} \stackrel{+2}{\underbrace{\overline{1}}_{-2}} X_{3}$ $d_{t}\boldsymbol{z} = \mathbb{W}\boldsymbol{z}$ Closed dynamics

 $= \mathbb{W}\boldsymbol{z}$ dynamics $\overset{\mathbb{W}\boldsymbol{\pi}^{\mathrm{eq}} = 0}{\lambda(\mathbb{W}) \leq 0} \quad \boldsymbol{z}(t) \to \boldsymbol{z}_{\mathrm{eq}} = L^m(0)\boldsymbol{\pi}^{\mathrm{eq}}$

$$\begin{array}{ccc} \mathbb{D} \neq \mathbb{O} \\ \hline & & \\$$

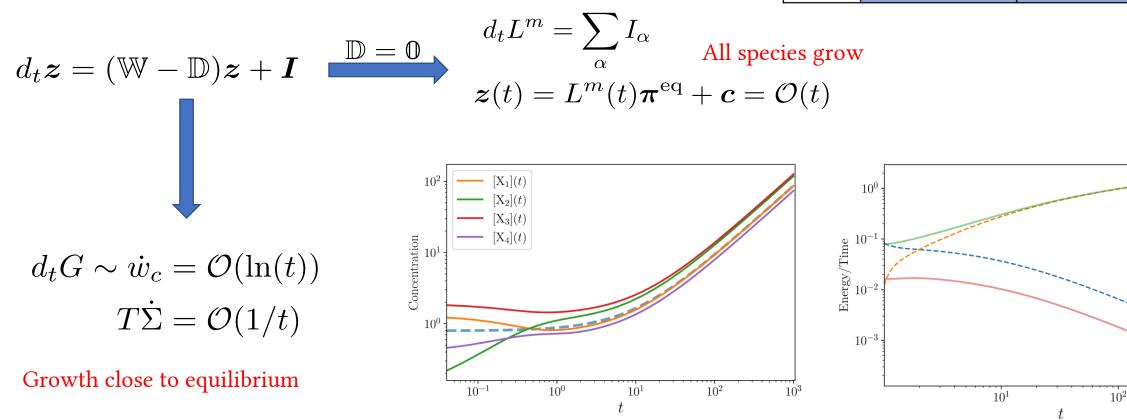
Unimolecular CRNs

		Flux Control	Mixed Control	Conc. Control
Dyn Line	1amically ear	Growth	No Growth	No Growth
Dyn Non	namically nlinear	Growth	Growth	No Growth*

--- $T\dot{\Sigma}$

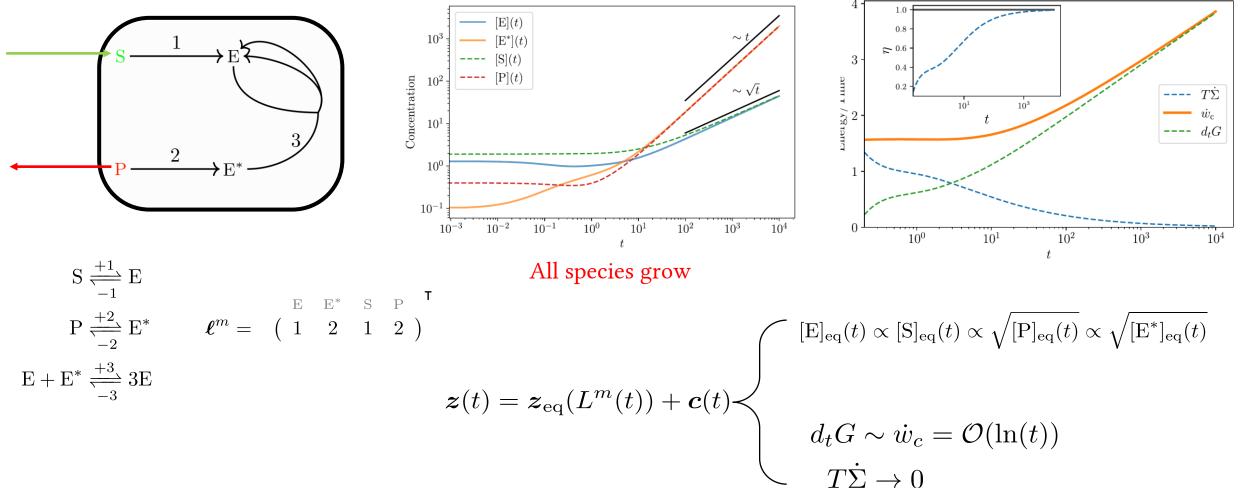
 $d_t G \ \dot{w}_{
m m} \ \dot{w}_{
m nc}$

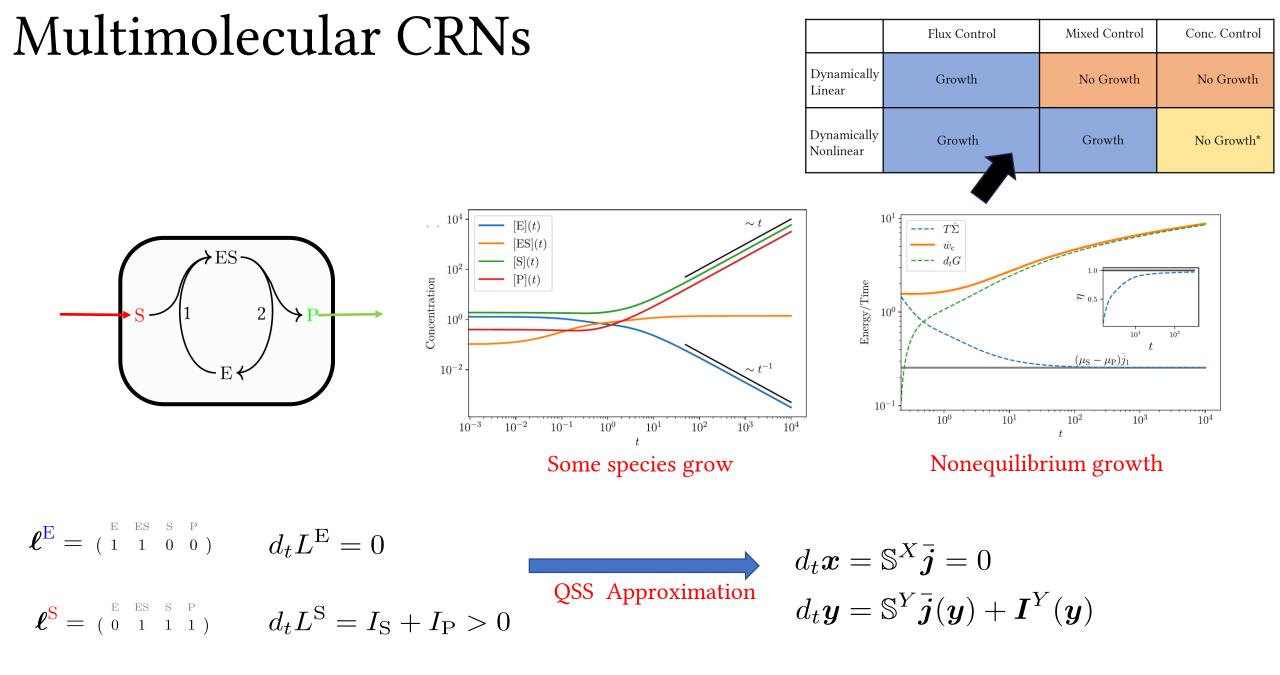
 10^{3}

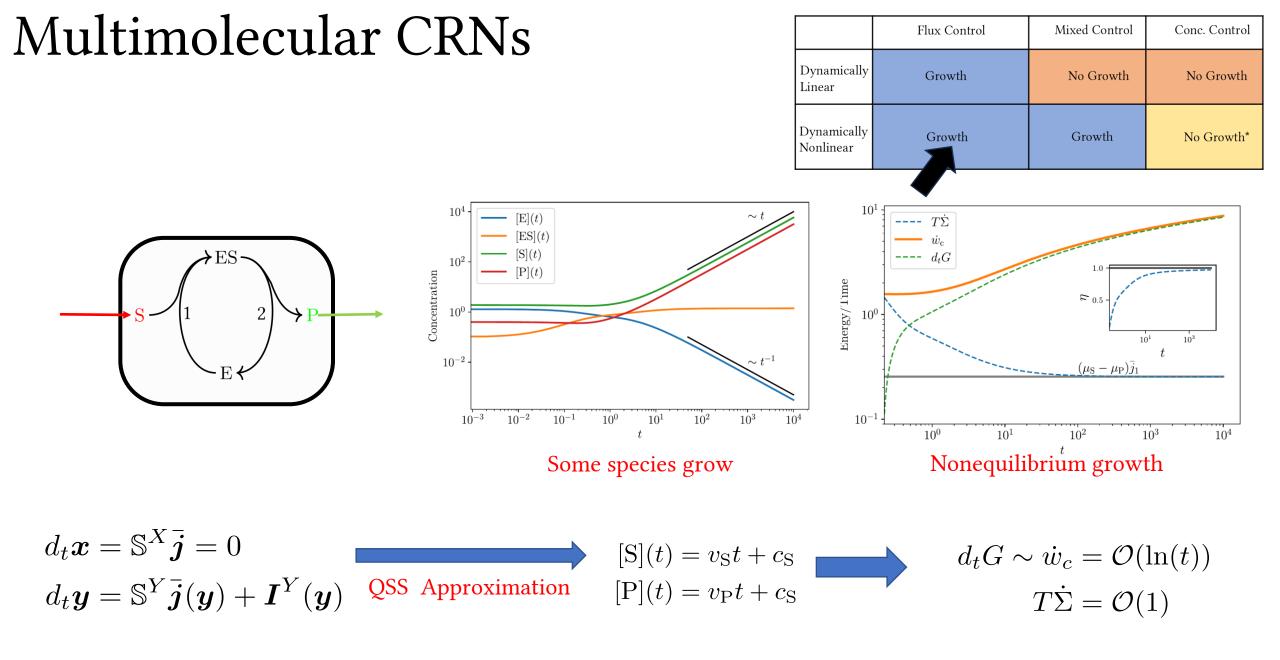


	Flux Control	Mixed Control	Conc. Control
Dynamically Linear	Growth	No Growth	No Growth
Dynamically Nonlinear	Growth	Growth	No Growth*

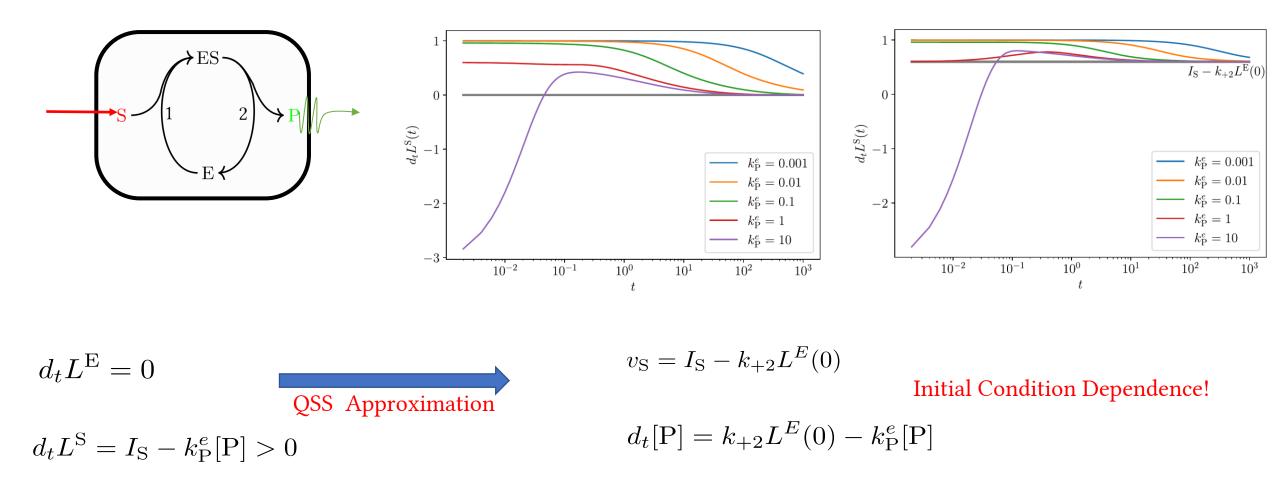
Nonlinear Growth close to equilibrium



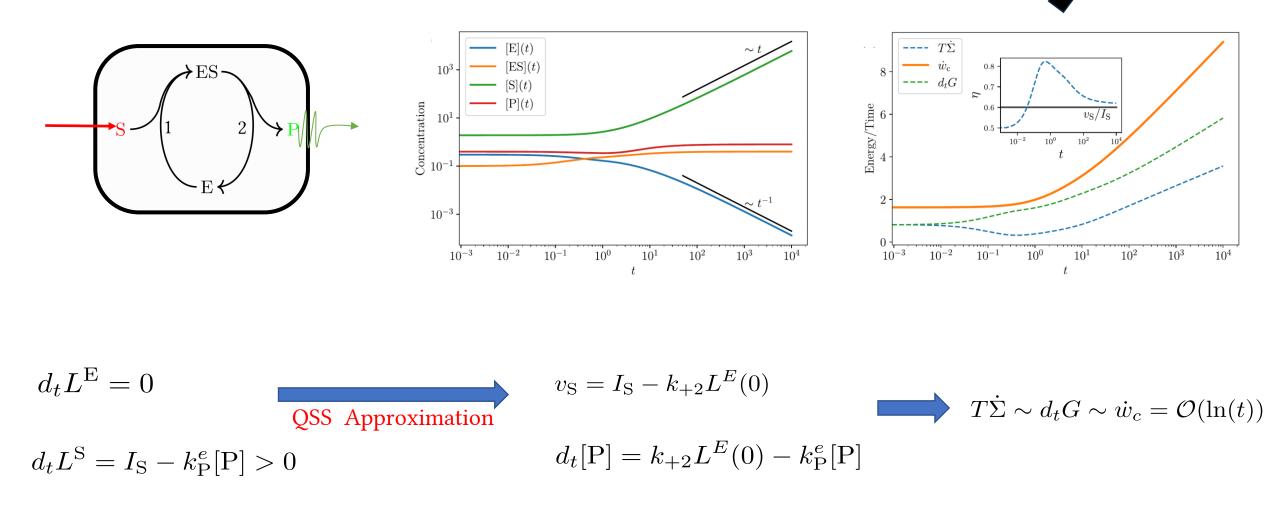


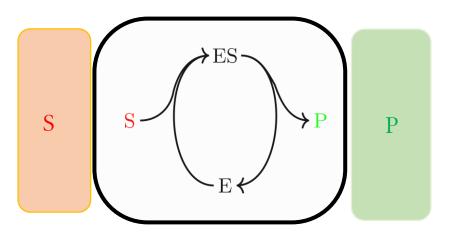


	Flux Control	Mixed Control	Conc. Control
Dynamically Linear	Growth	No Growth	No Growth
Dynamically Nonlinear	Growth	Growth	No Growth*



	Flux Control	Mixed Control	Conc. Control
Dynamically Linear	Growth	No Growth	No Growth
Dynamically Nonlinear	Growth	Growth	No Growth*





	Flux Control	Mixed Control	Conc. Control
Dynamically Linear	Growth	No Growth	No Growth
Dynamically Nonlinear	Growth	Growth	No Growth*

Growth $\iff L^m(t) \to \infty \iff G(z) \to \infty$

Known results: Boundedness of weakly reversible single linkage class CRNs J Math Chem 49 2275 (2011) Idea: Gibbs Energy as a quasi-Lyapunov function

Numerical evidence!

Summary of Results

- Asymptotic Growth needs influx of species
- Topology of the CRN determines whether the growth is equilibrium/nonequilibrium

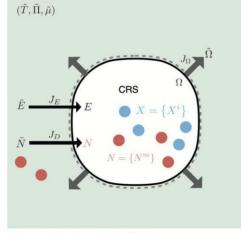
	Flux Control	Mixed Control	Conc. Control
Unimolecular CRNs	$egin{aligned} oldsymbol{z}(t) &= \mathcal{O}(t) \ T\dot{\Sigma} &= \mathcal{O}(1/t) \ \eta & ightarrow 1 \end{aligned}$	$oldsymbol{z}(t)=\mathcal{O}(1)$	$oldsymbol{z}(t)=\mathcal{O}(1)$
Pseudo- unimolecular CRNs	$egin{aligned} oldsymbol{z}(t) &= \mathcal{O}(t) \ T\dot{\Sigma} &= \mathcal{O}(t) \ \eta & ightarrow 0 \end{aligned}$	$oldsymbol{z}(t)=\mathcal{O}(1)$	$oldsymbol{z}(t)=\mathcal{O}(1)$
Multimolecular CRNs	$0 \le \eta \le 1$	$0 \le \eta \le 1$	

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- Prospective directions

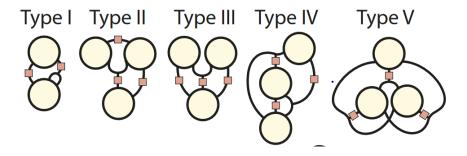
Prospective Directions

- Systematic analysis of multimolecular CRNs
- Transient Growth
- Growth with volume



Chemical Reaction System (CRS)

Phys. Rev. Research 4, 033191(2022)



PNAS, 117 , 25230 (2020) J Math Biol. 2022 Sep 7;85(3):26 Phys. Rev. E 100, 022414 (2019)

THANK YOU!